

LETTER TO THE EDITORS

Temperature Dependence of the Thermal Diffusion Length in Water

The temperature dependence of the thermal diffusion length in water has been given by Fermi (1) to be, $L = 2.64 + 0.0061T$ (L in cm and T in °C). This expression is based on the unpublished work of Wilson, Bragdon, and Kanner, who made several measurements in the temperature range 27.2–93°C. For a Maxwellian neutron spectrum, the analysis of these results reveals that the assumption of a $1/v$ energy dependence of the hydrogen transport cross section gives an excellent fit to the data.

The recent measurements of Von Dardel and Sjöstrand (2) yield a thermal diffusion length of 2.725 ± 0.03 cm at 22°C. This result is in very good agreement with the measurement of De Juren and Rosenwasser (3) corrected for the effect of the buckling on the neutron temperature (2). Using this value of the thermal diffusion length and assuming the $1/v$ dependence, the transport cross section of hydrogen in water at a velocity of 2200 meters/sec is 37.14 barns.

The diffusion length at any other temperature can be found from the value at 22°C using the following expression:

$$L = \frac{2.719}{\rho} \left[\frac{T + 273.1}{295.1} \right]^{\frac{1}{2}} \left[\frac{1.0614}{1 + 0.0614 \left[\frac{T + 273.1}{295.1} \right]^{\frac{1}{2}}} \right]^{\frac{1}{2}}$$

where ρ is the water density and T the temperature in °C. This expression agrees very well with the results of Wilson, Bragdon, and Kanner, as well as with the results of Antonov *et al.* (4) who made measurements at 23°C and 80°C. Using the above expression for the diffusion length and approximating it by a linear temperature dependence in the 10–30°C and 30–90°C temperature intervals, the temperature coefficient in the range 30–90°C is 0.0059 cm/°C, while in the range 10–30°C the value is 0.0049 cm/°C. This latter value is in somewhat better agreement than the Fermi prescription with the value of 0.003 cm/°C reported by Von Dardel and Sjöstrand over a similar temperature range.

REFERENCES

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